

Qualitative Phytochemical Screening of MenthaarvensisL.

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ABSTRACT:

Phytochemicals plant non-nutritive are chemicals.Plant metabolites represent enormous chemical diversity with largely unexplored biological activities. Our extensive knowledge on the chemistry and pharmacology of some secondary metabolites has led to their use in a range of medical applications. Thin Layer Chromatography is a common analytical technique widely used for the analysis of phyto-constituents in plant extracts. In the present study was to investigate the presence of various phytochemical from different solvents (methanol, ethanol, ethyl acetate, petroleum ether, hexane and aqueous) extracts of Menthaarvensis L.Among them methanol, ethanol solvent extracts were found with richsecondary metabolites (flavonoids, glycosides, phenols, terpenoids, alkaloids) due to highest number of various metabolites compounds.

KEYWORDS: Menthaarvensis L, In Vitro Regeneration, Phytochemicals, Solvents, Extraction, TLC

I. INTRODUCTION

Since ancient times, people have been exploring the nature particularly plants in search of new drugs. The medicinal plants are useful for healing as well as for curing of human diseases because of the presence of phytochemical constituents (Nostroet al., 2000). Phytochemicals are fascinating chemical molecules, very useful and of great importance in nature, as well as highly diversified in structures, properties, uses, chemistry etc.MenthaarvensisL. is well known important medicinal and aromatic plant widely used in several indigenous systems of medicine as various therapeutic powers viz. analgesic, anesthetic, antiseptic, astringent, carminative, decongestant, expectorant, nervier, stimulant, inflammatory disease, ulcer and stomach problems (Campbell et al., 1973; Blumenthal, 1998; Jamal et al., 2006). In India, mint is used to tone the stomach, stimulate the mind and body, rid the intestines of gas, and

relieve muscle spasms (Pandey, 2003). The Commission E approved internal use of mint oil for flatulence, functional gastrointestinal and gallbladder disorders, catarrhs of the upper respiratory tract, and external use for myalgia and neuralgic ailments. Menthol crystals are used in different pharmaceutical products and cosmetics as antiseptic, stimulant and inhibitor. It gives minty flavour to various food products. It is also used in oral products e.g. tooth paste and mouth fresheners due to its physiological cooling effect. Extensively used as fragrance component in soaps, detergents, cosmetics and perfumes, toothpastes, and industrial fragrances (Alviet al., 2001).

II. MATERIALS AND METHODS

In vitro plants were removed from MS supplemented with 2.0 mg/l BAP + 0.4 mg/l NAA standardized cultured medium (Bariya and Pandya, 2014). Whole plant material was washedunder the running tap water. 10 gm of plant material (shade dried powder) was extracted separately with 100 ml of each solvents i. e. water, ethanol, methanol, ethyl acetate, petroleum ether and hexane and allowed to stand for 24 hours soaked in air tight Erlenmeyer flask. Later it filtered through a whatman filter paper no. 1 (Souriet al,2008; Khan and Nasreen, 2010). The filtrate was evaporated for drying to yield a dark-residue and % yield of extracts were calculated. Each sample was then transferred to glass vials and kept in refrigerator at 4°C for their future use in phytochemical analysis.

Distillation of essential oil-An air-dried powdered sample (100 gm) was subjected to hydro distillationfor 3 h using a Clevenger-type apparatusto produce the essential oil. The obtained essential oil was dried over anhydrous sodium sulphate and stored at 4 °C for further experiments. Hydrosols are the condensate water coproduced during the Hydro distillation of plant material.



Phytochemical Screening

The qualitative phytochemical screening of the plant extracts using carried out by testing of different class of compounds using standard methods (Harborne, 2005; Raman, 2006). In TLC screeningChromatogaphyplates were prepared by spreading slurry of Silica Gel G in distilled water (1:2) uniformly over clean glass plates (20 cm.) The layer (thickness 2 mm) was allowed to dry and activated by heating in an oven at 110 °C for one hour. Samples were applied to adsorbent surface at 10 mm edge using a capillary tube and developed in glass chamber preciously saturated with the vapours of standardized solvent system. after the development plates left to dry for about 10 minutes, then viewed under UV fluorescence light at 254 and 366 nm wavelength and finally sprayed with the required detection reagent (dragendorf reagentsulphuric alkaloids. vanillin acid-terpenoids, saponins, Ferric ferrocynide- phenols, ammonium vapour-flavanoids after dry the plate heated at 110° C for 5-10 min. Then plate was evaluated for chromatographic measurement to determine the compounds present.

III. RESULTS AND DISCCUSION

Phytochemical results revealed the presence of various bioactive secondary metabolites in the different solvent extracts (Table-1). The maximum numbers of secondary metabolites were observed in ethanolic and methanolic extracts which were rich in flavonoids, phenols, glycosides, alkaloids and terpenoids. Ethyl acetate extract shown the presence of high amounts of phenols and flavonoids. Petroleum ether extract shown highest amount of terpenoids, very less of alkaloids and absence of glycosides. Hydrosol sample (aroma) shown high content of flavonoids, terpenoids, saponins and glycosides. The similar findings were also reported by John et al., (2012), Suresh (2012) and Rachel and MeeraBai (2011). In addition Singh et al. (2011), Naidu et al. (2012) who observed in essential oil contain most of the phytoconstituents including flavonoids, saponins, cardiac glycosides, reducing sugars and steroids, alkaloids.

Phyto constitue nts	Aqu eous extra ct	Methano l extract	Ethanol extract	Ethyl acetate extract	Petroleum ether extract	Hexane extract	Essential oil	Hydrosol
Alkaloids								
1.	+	++	+++	+	+	-	+	+
Mayer's	+	++	++	+	+	+	+	+
test								
2.Dragen	+	+	+	-	+	+	-	+
dore's								
test								
3.Wagner								
's test								
Flavonoids	5							
1.	++	+++	+++	++	++	++	++	++
Alkaline								
reagent	+	++	++	++	+	+	+	++
2. Fecl_3								
test								
Phenolic c	ompoun	ds		-				
1. Lead	+	++	+++	+++	++	+	++	+
acetate								
test	++	++	++	++	++	+	++	+
2. $Fecl_3$								
test								
Terpenoid	s							



1.Liberm	-	++	++	-	+++	+++	++	++
ann test	-	+	+	+	++	++	++	++
2.								
Salkowas								
kis test								
Glycosides	;							
1. Keller-	+	+	+	+	-	-	+	++
killiani								
test	+	+	+	-	-	-	+	+
2.								
Borntrag								
er's								
test(athra								
quinone								
glycoside								
s)								
Saponins								
1. Froth	+	++	++	+	+	+	++	++
test								

TABLE-2 FLAVANOIDSOLVENT SYSTEM

			No. of Ban	ds				
S. No.	Solvent system (Flavanoids)		Methanol Extract	Ethanol Extract	Ethyl acetate Extract	Petroleum ether Extract	Hexane Extract	Aqueous Extract
1.	ethyl acetate: formic acid : glacial acetic acid : water (100:11:11:20)	Treatment with ammonia	3	3	2	2	2	1
2	con. HCL: acetic acid: water (3:30:10)		2	2	1	1	1	2
3	butanol: acetic acid: water (4:1:5)	vapour	3	3	2	1	2	3
4	Phenol: water (3:1)		3	4	3	3	2	3

TABLE-3 ALKALOID SOLVENT SYSTEM

			No. of Bands							
S. No.	Solvent system (Alkaloids)		Methanol Extract	Ethanol Extract	Ethyl acetate Extract	Petroleum ether Extract	Hexane Extract	Aqueo us Extra ct		
	cyclohexane:	Before Spray	5	5	2	4	3	4		
1.	ethanol: diethylamine(80:10:10)	After Spray(Drag.+ H ₂ So ₄)	7	8	3	7	5	8		
		H_2So_4	11	10	6	9	8	13		



2	toluene: ethyl acetate: diethyl amine(70:20:1 0)	After Spray(Drag.+ H ₂ So ₄)	4	5	-	3	-	6
	butanol:	Before Spray	2	2	-	1	1	3
3	acid: water (40:40:10)	After Spray(Drag.+ H ₂ So ₄)	3	3	1	1	1	5

(Drag.=Dragendoeff's reagent)

TABLE-4GLYCOSIDES SOLVENT SYSTEM

			No. of Bands						
S. No ·	Solvent system (Glycosides)	After Spray(An isaldehyd	Methano l Extract	Ethano l Extrac t	Ethyl acetate Extrac t	Petroleu m ether Extract	Hexan e Extrac t	Aqueou s Extract	
1.	ethyl acetate: methanol: ethanol: water(81:11:4: 8)	e sulphuric acid reagent)	8	9	3	2	2	4	

TABLE-5PHENOLS SOLVENT SYSTEM

	Salvant		No. of Ban	ds						
S. No.	solvent system (Phenols)		Methanol Extract	Ethanol Extract	Ethyl acetate Extract	Petroleum ether Extract	Hexane Extract	Aqueous Extract		
1.	benzene : ethyl acetate (11:9)	Ferric ferrocynide	9	9	4	5	2	6		
2	chloroform: acetic acid (9:1)	reagent	7	8	6	6	7	8		

TABLE-6 TERPENOID SOLVENT SYSTEM

			No. of Ban	ds				
S. No.	Solvent system (terpenoids)		Methanol Extract	Ethanol Extract	Ethyl acetate Extract	Petroleum ether Extract	Hexane Extract	Aqueous Extract
1.	Dichloromethane		9	10	9	8	7	5
2	toluene: chloroform (1:1)	Vanillin sulphuric	10	12	8	6	4	5
3	toluene: ethyl acetate (97:3)	acid reagent	8	10	8	5	6	3
4	chloroform: benzene (1:1)		6	9	4	6	8	3
5	chloroform: ethyl acetate: ammonia		15	16	7	15	15	1



	(97:2.5:05)						
6	chloroform: methanol (90:10)	12	14	13	15	13	4

The results of TLC screening of ethanolic and methanolic extracts exposed with maximum number of bands in all the solvent systems. Ethyl acetate extract revealed maximum number of bands presence in phenols and flavonoid solvent system followed by petroleum ether extracts detected 15 bands in terpenoid solvent system, aqueous extracts found with highest number of bands in phenol and alkaloid solvent system.

In study of different solvent systems, chloroform: ethyl acetate: ammonia showing maximum number of bands followed by cyclohexane: ethanol: diethylamine (terpenoids), benzene: ethyl acetate (phenol), Ethyl acetate: methanol: ethanol: water (glycosides) and Phenol: water (flavonoids). When a new drug is to be discovered, qualitative phytochemical analysis is a very important step as it gives information about the presence of secondary metabolite in plant extracts which having a clinical significance. TLC screening of all extracts gives a remarkable result that directing towards the presence of number of phytochemicals. Various phytochemicals have different R_f values in different solvent system. This variation in R_f values provide a very important clue in understanding of polarity of compound.

IV. CONCLUSION

In present investigation, methanol and ethanol extracts of Menthaarvensis L. showed the high and positive response than other solvent extracts. Qualitative phytochemical screening is necessary for determination of metabolites occurring in plant. For the further study of metabolites, secondary screening was done by Thin layer chromatography technique using different solvent system. TLC provides a very important clue in understanding of compounds polarity and also helps in selection of appropriate solvent system for separation of pure compounds.

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